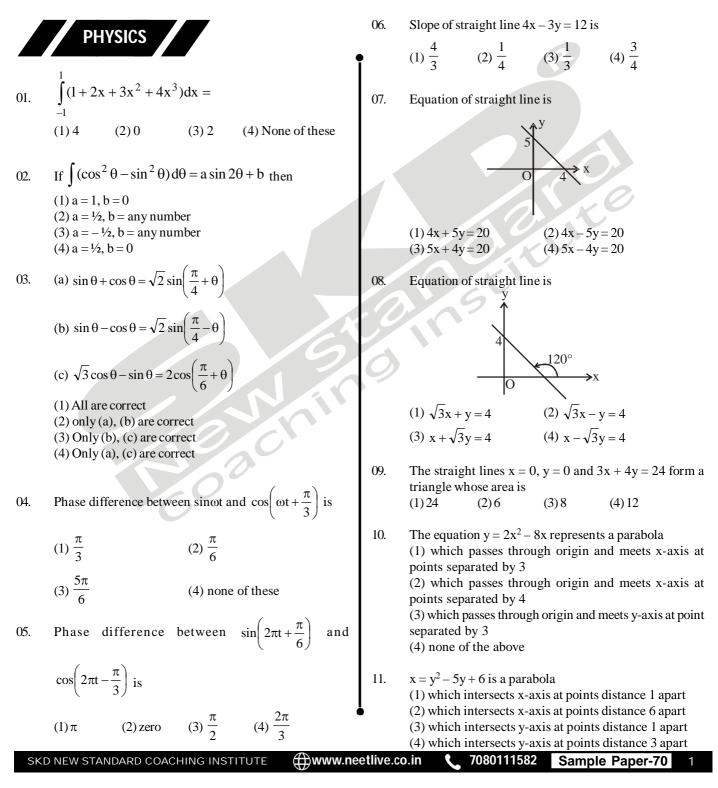


SAMPLE PAPER - 70

Time : 1 : 15 Hr.

Question : 60



12. If $N = N_0 e^{-\lambda t}$, the terms are having as usual meaning.

Find out time t when $N = \frac{N_0}{e}$

(1).69
$$\lambda$$
 (2) λ (3) $\frac{.69}{\lambda}$ (4) $\frac{1}{\lambda}$

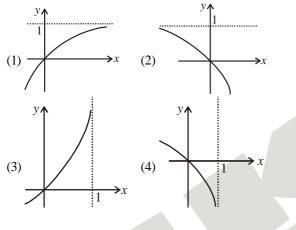
13. If x is numerically much less than 1, $\frac{(1-x)^{14}}{(1+x)^6}$ is

approximately equal to (1) 1+8x (2) 1-8x (3) 1+20x (4) 1-20x

14. If $2^x = 5$ and $\log_{10} 2 = 0.3$ then x =

(1)
$$\frac{7}{3}$$
 (2) $\frac{3}{7}$ (3) $\frac{5}{2}$ (4) none of these

15. Graph of $y = 1 - e^{-x}$ is



CHEMISTRY

- 16. Four one litre flasks are separately filled with the gases H_2 , He, O_2 and O_3 at the same temperature and pressure. The ratio of total number of atoms of these gases present in different flask would be: (1) 1 : 1 : 1 : 1 (2) 1 : 2 : 2 : 3 (3) 2 : 1 : 2 : 3 (4) 3 : 2 : 2 : 1
- 17. In a gaseous reaction of the type $aA + bB \longrightarrow cC + dD$, which is wrong?

(1) a litre of A combines with b litre of B at same P& T to give C and D (2) a male of A combines with b male of B to give C and D

(2) a mole of A combines with b mole of B to give C and D
(3) a g of A combines with b g of B to give C and D
(4) a molecules of A combines with b molecules of B to give C and D

18. 1 amu is equal to

(1)
$$\frac{1}{12}$$
 of C - 12 (2) $\frac{1}{14}$ of O - 16
(3) 1 g of H₂ (4) 1.6×10^{-23} kg

- 19. The simplest formula of a compound containing 50% of element X (atomic mass 10) and 50% of element Y (atomic mass 20) is
 (1)XY
 (2)X₂Y
 (3)XY₃
 (4)X₂Y₃
- 20. The atomic weights of two elements A and B are 40 and 80 respectively. If x g of A contains y atoms, how many atoms are present in 2x g of B?

(1)
$$\frac{y}{2}$$
 (2) $\frac{y}{4}$ (3) y (4) 2y

21. The molecular weight of O_2 and SO_2 are 32 and 64 respectively. At 15 °C and 150 mm of Hg pressure, one litre of O_2 contains 'N' molecules. The number of molecules in two litres of SO_2 under the same conditions of temperature and pressure will be:

(1)
$$\frac{N}{2}$$
 (2) N (3) 4N (4) 2N

- 22. The total number of electrons in one molecule of carbon dioxide is
 (1) 22 e (2) 22 mol e (3) 66 e (4) 66 mol e
- What volume of CO₂ will be liberated at NTP, if 12 g of carbon is burnt in excess of oxygen?
 (1)11.2 L
 (2)22.4 L
 (3)2.24 L
 (4)1.12 L

24. How many atoms are contained in one mole of sucrose $(C_{12}H_{22}O_{11})$ (1) 45 × 6.02 × 10²³ atoms/mole (2) 5 × 6.02 × 10²³ atoms/mole (3) 5 × 6.02 × 10²⁰ atoms/mole (4) None of these

- 25. Number of mole in 1 m³ gas at NTP are: (1)44.6 (2)40.6 (3)42.6 (4)48.6
- 26. A gas mixture contains O_2 and N_2 in the ratio of 1: 4 by weight. The ratio of their number of molecules is (1)1:8 (2)1:4 (3)3:16 (4)7:32
- 27. If the volume of x molecules of H_2 gas at STP is 5L. What will be the volume of x molecules of O_2 gas at STP?

(1)
$$\frac{5}{80}$$
L (2) $\frac{16}{5}$ L (3) $\frac{5}{16}$ L (4) 5 L

28. Ten mole of P_4 molecules contain: (1) 1 molecule (2) 24.088 × 10²⁴ atoms

(3)
$$\frac{1}{4} \times 6.022 \times 10^{23}$$
 atoms
(4) 24.088 × 10²³ atoms

29. When 22.4 litres of $H_2(g)$ is mexed with 11.2 litres of $Cl_2(g)$ each at STP, the moles of HCl(g) formed is equal to: (1) 5.0 mol of HCl(g) (2) 1.5 mol of HCl(g)(3) 1.0 mol of HCl(g) (4) 2.0 mol of HCl(g) Persons are medically considered to have lead poisoning if they have a concentration of greater than 10 μg of lead per decilitre of blood. Concentration in parts per billion is (1) 1000 (2) 100 (3) 10 (4) 1



31. When molecules move across in a membrane independent of other molecule through carrier protein, then the process is known as
(1) Symport (2) Antiport

(4) All of these

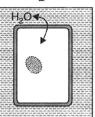
- (3) Uniport
- 32. How can we get egg membrane?
 (1) Remove yolk and albumin through a small hole at one end of the egg
 (2) Planting the line with the Will for for all the state of the end of the egg
 - (2) Place the shell in dilute HCl for few hours
 - (3) Both (1) and (2)
 - (4) None of these
- 33. Deplasmolysis occurs in a cell when it is placed in (1) Hypotonic solution(2) Hypertonic solution
 - (3) Isotonic solution
 - (4) Buffer solution

34. In flowering plants, we need to transport

- (1) Water and mineral nutrients
- (2) Organic nutrients
- (3) Plant growth regulators
- (4) All of these
- 35. Transport over longer distance is known as
 (1) Translocation
 (2) Transformation
 (3) Transduction
 (4) Diffusion
- 36. Diffusion can occur between

 (1) One part of cell to other part
 (2) Cell to cell
 (3) Intercellular space to outside of leaf
 (4) All of these
- 37. Which of the following process undergoes saturation?
 (1) Facilitated diffusion (2) Active transport
 (3) Simple diffusion (4) Both (1) and (2)
- 38. The two main components which determine water potential are
 (1) Solute potential
 (2) Pressure potential
 (3) Matrix potential
 (4) Both (1) and (2)
- 39. Dry seeds when placed in water swells due to
 (1) Imbibition
 (2) Absorption
 (3) Diffusion
 (4) Adsorption

- 40. A cell increases in volume if the external medium is (1) Hypotonic (2) Hypertonic (3) Isotonic (4) None of these
- 41. A cell is plasmolyzed after being kept in a hypertonic solution. What will be present between the cell wall and plasmalemma?
 (1) Isotonic solution
 (2) Hypertonic solution
 (3) Air
 (4) Hypotonic solution
- 42. Diffusion rate is affected by
 (1) Concentration gradient
 (2) Membrane permeability through which it occurs
 (3) Pressure and temperature
 (4) All of these
- 43. Where porin proteins are present?
 - (1) Outer membrane of plastid
 - (2) Inner membrane of mitochondria
 - (3) Inner membrane of some bacteria
 - (4) All of these
- 44. Which of the following has maximum water potential? (1) 1 M of NaCl (2) 0.5 M of glucose (3) Pure water (4) 0.001 M of HCl
- 45. Identify the figure and select the correct statement.



(1) The movement of water occurred across the membrane moving from an area of high water potential to an area of lower water potential

(2) Water flows into the cell and out of the cell and are in equilibrium

(3) Water diffuses into the cell causing the cytoplasm to build up a pressure against the wall

(4) None of the above



46. What is serum?

(1) (Blood) - (Plasma) (2) (Pl_{a}) (Pl_{b}) ($Pl_$

- (2) (Blood) (Plasma + RBC)
- (3) (Plasma) -(Clotting factor)
- (4) (Plasma) (WBC)
- 47. The main osmotic protein of blood is
 (1) Albumin (2) Globulin
 (3) Fibrinogen (4) Thromboplastin

3

48.	The shape of RBC in m (1) Oval (3) Biconcave	ammal is (2) Bicon vex (4) Flattened	57.	Chordae tendineae are (1) Joints (3) Ventricles of heart	found in (2) Atria of heart (4) Ventricles of brain
49.	. ,	f the following blood cell leads of	58.	SA node is located in	(+) ventretes of brain
	bleeding? (1) Thrombocytes (3) Monocytes	(2) Neutrophils (4) RBCs		 (1) Upper left corner of f (2) Lower left corner of (3) Lower right corner of (4) Upper right corner of 	left atrium f left atrium.
50.	The cell involved in inf (1) RBCs (3) Basophils	flammatory reaction is (2) Platelets (4) All of these	59.	Bundle of His is a grou (1)Ganglia	-
51.	Megakaryocytes are found in			(3) Muscular fibres (4) Connective tissue	
51.	(1) Lungs	(2) Liver	60.	During ventricular syst	ole, oxygenated blood is pumped
	(3) Bone marrow	(4) Spleen	00.	into the	nated blood is pumped into the
52.	Which of the following cells are responsible for immune response of the body?			pulmonary artery. (2) Pulmonary artery and deoxygenated blood is pumped	
	(1) T lymphocyte(3) Both (1) and (2)	(2) B lymphocyte(4) Astrocytes		into the artery.(3) Aorta and deoxygpulmonary vein.	genated blood is pumped into
53.	Find the correct decending order to percentage proportion of leucocytes in human blood.			(4) Dulmonomy woin and doorwoon stad blood is numbed	
	 Neutrophils → Basophils → Lymphocytes → Acidophils (Eosinophils) → Monocytes Neutrophils → Monocytes → Lymphocytes → Acidophils → Basophils 			(4) Pulmonary vein and into pulmonary artery.	
	Acidophils \rightarrow Basoph				
	(4) Neutrophils \rightarrow Lymphocytes \rightarrow Mor	Acidophils \rightarrow Basophils \rightarrow nocytes		5	
54.	 Closed circulatory system is found in (1) Arthropod and chordates. 				
	(2) Molluscs and chordates.				
	(3) Amphibians and molluscs.				
	(4) Annelids and chordates.				
55.	The following diagram represents circulation in Deoxygenated				
	Heart	Gills			
		Body			
	↓	L			
	Deoxygenated (1) Fishes (2) Amphibians				
	(3) Birds	(4) Reptiles			
		· · · · · ·			
56.		valve is found between	l		
	 (1) Left attrium and left ventircle. (2) Right Atrium and right ventricle. 				
	(2) Right Atrium and ri (3) Right atrium and le				
	(4) Left atrium and righ				
	()				

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